

# CrO<sub>2</sub> thin film preparation

first results

# CrO<sub>2</sub>

- Chromium dioxide (CrO<sub>2</sub>) is a metallic ferromagnetic oxide.
- The chromium ions are in the Cr<sup>+4</sup> state with the electronic configuration [Ar]3d<sup>2</sup> with a magnetic moment of 2μ<sub>B</sub> per ion. Theoretical calculations have predicted CrO<sub>2</sub> to be half metallic, with almost complete spin polarization at the Fermi level.

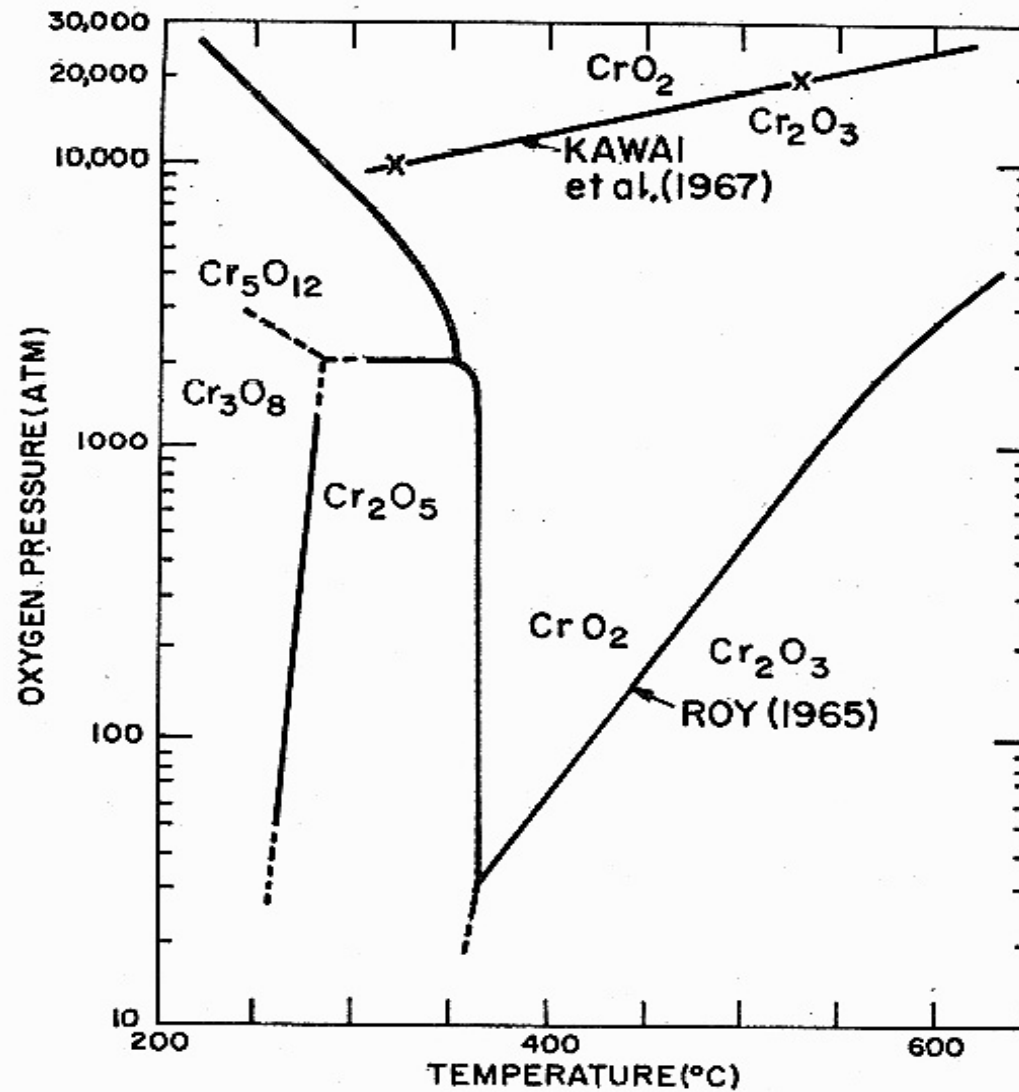
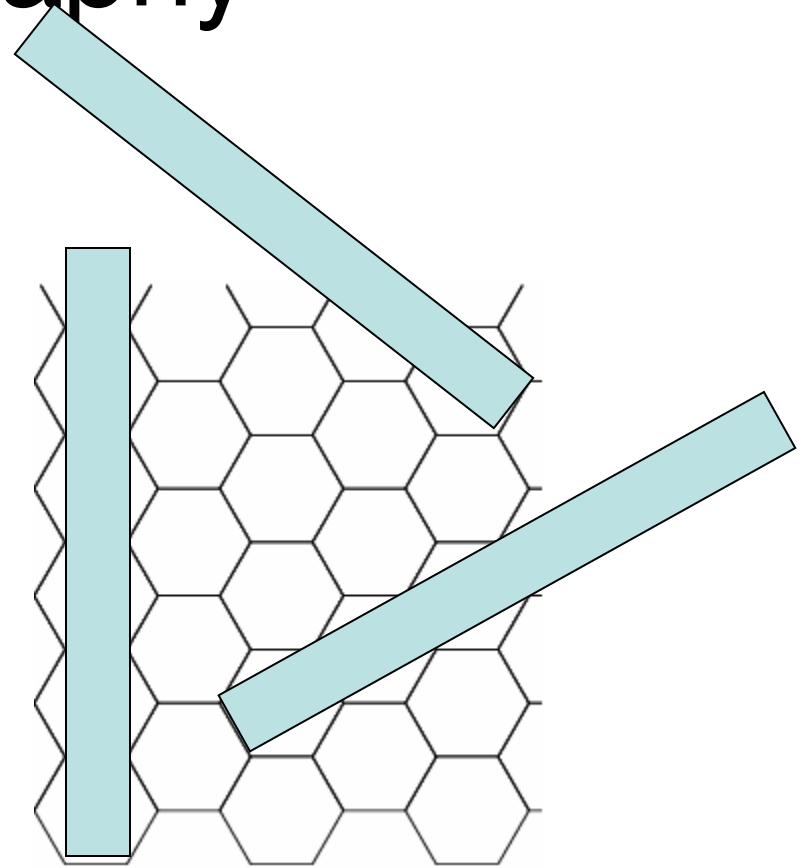


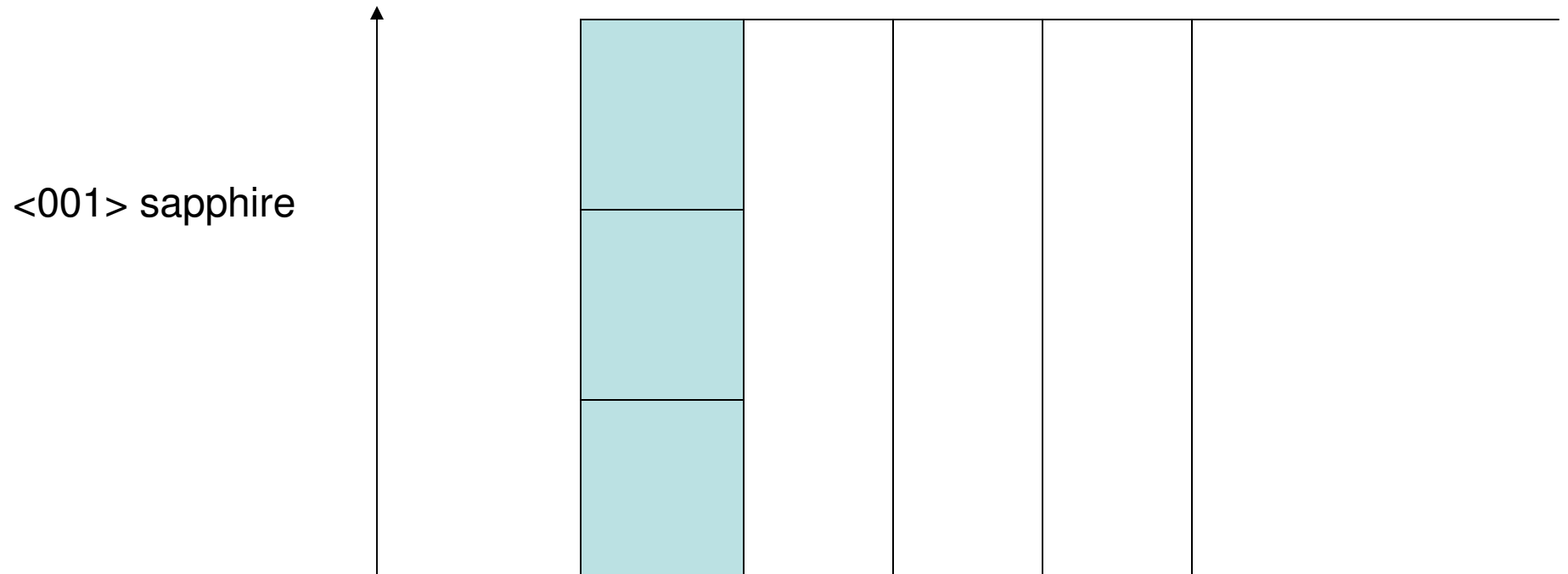
Figure 1. Phase diagram of the chromium-oxygen system

# crystallography

- CrO<sub>2</sub> tetragonal  $a=0.442$  nm
- $c=2.912$  nm
- Sapphire hexagonal  $a=0.448$  nm
- $c=1.304$  nm
- $\langle 1\ 0\ 0 \rangle$  of CrO<sub>2</sub> aligned with  $\langle 0\ 0\ 1 \rangle$  of sapphire

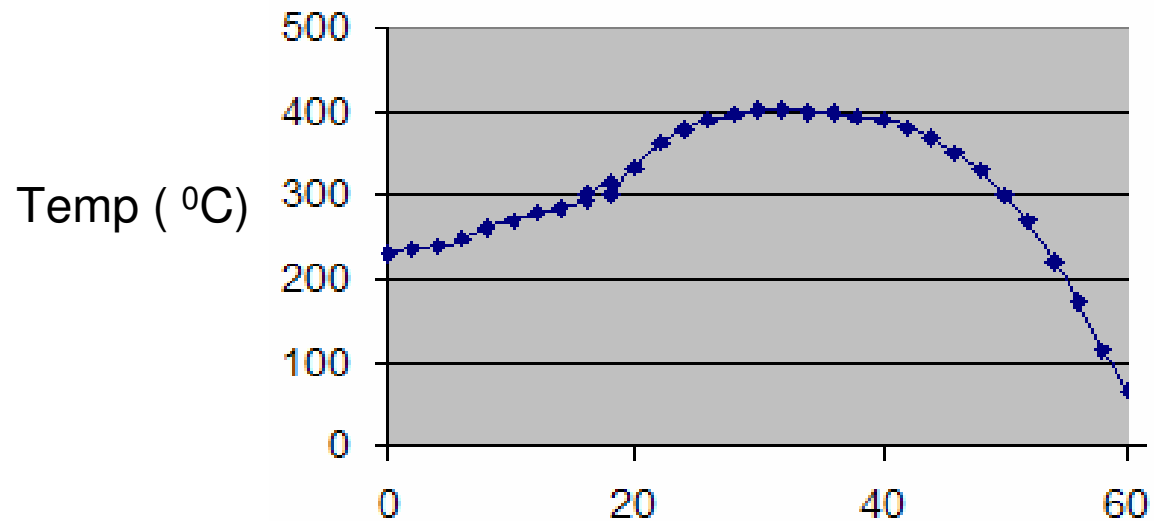
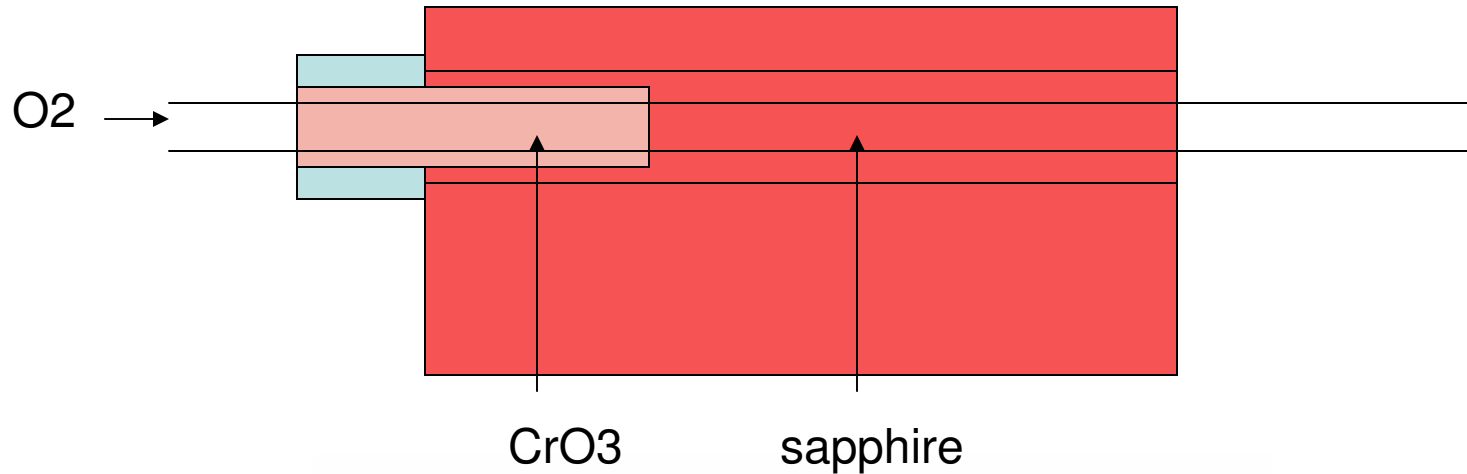


# Sideview CrO2 on sapphire

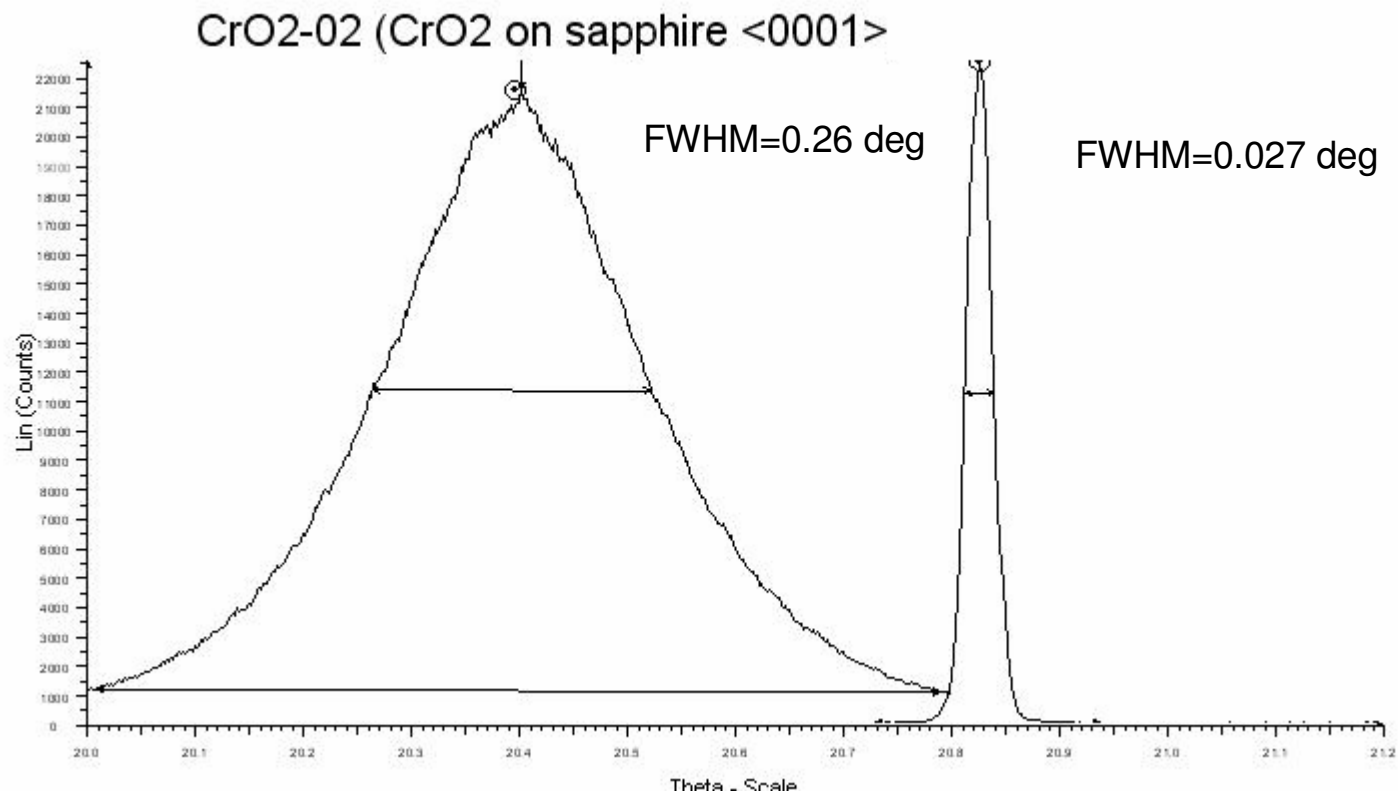


# Preparation

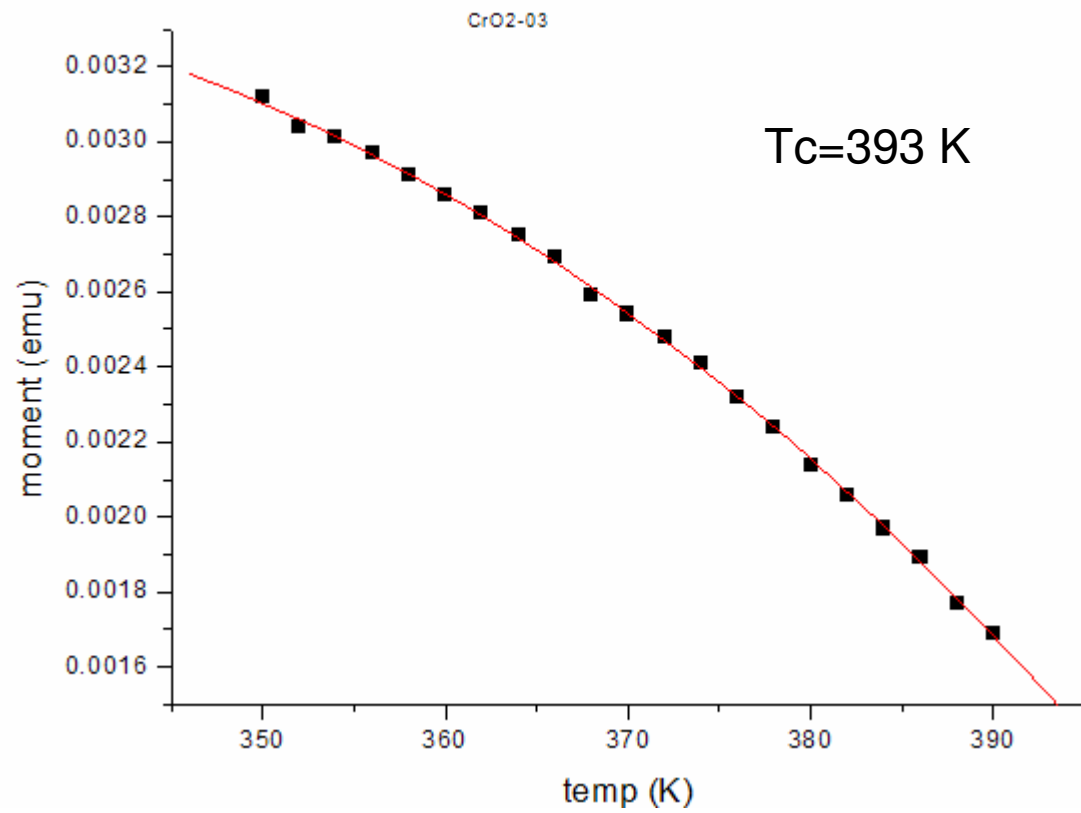
decomposition reaction  $\text{CrO}_3 \rightarrow \text{CrO}_2 + \frac{1}{2} \text{O}_2$



# Rocking curves

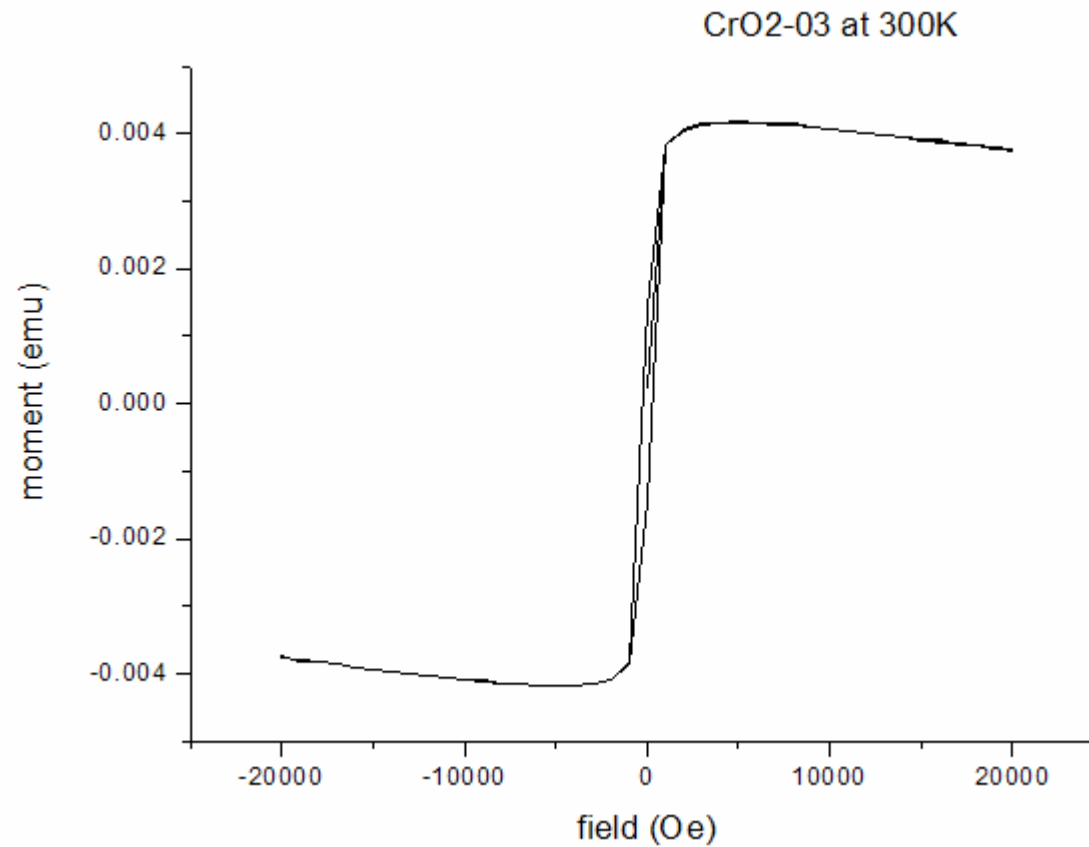


# M vs T

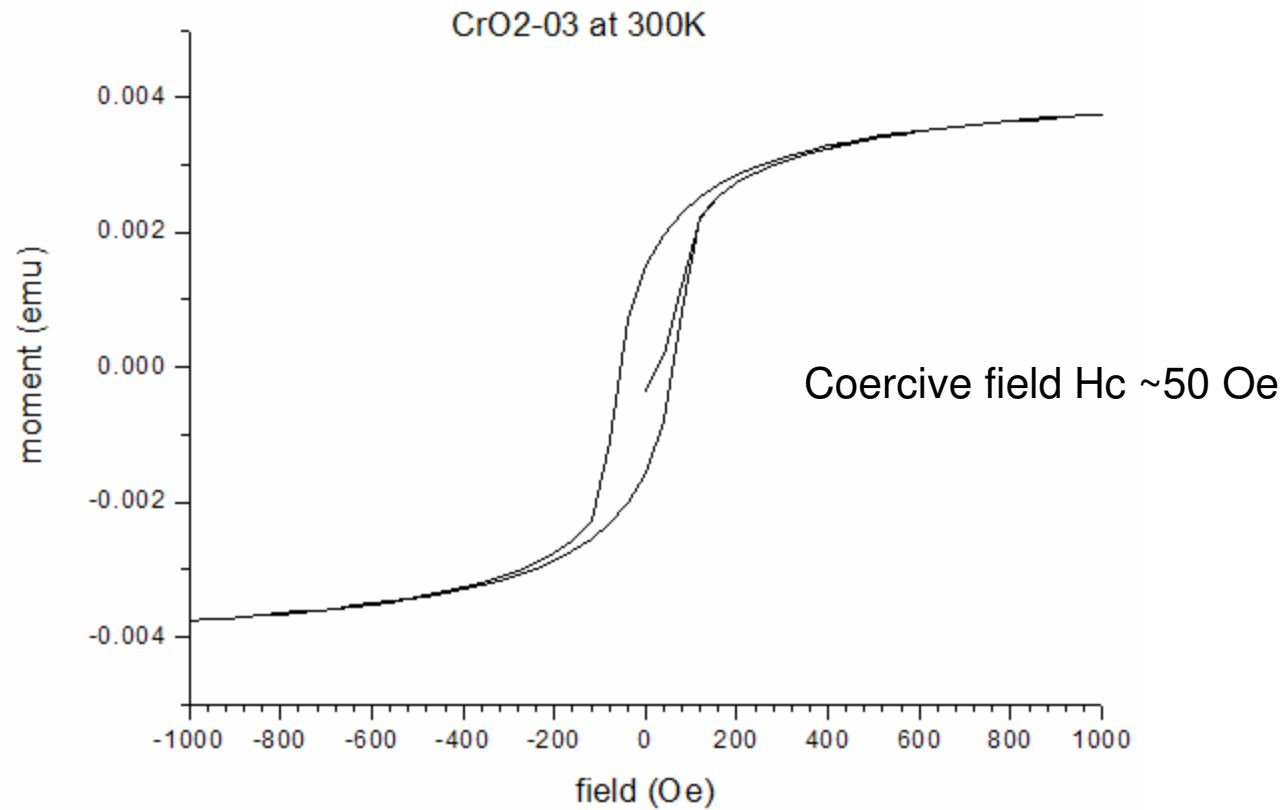




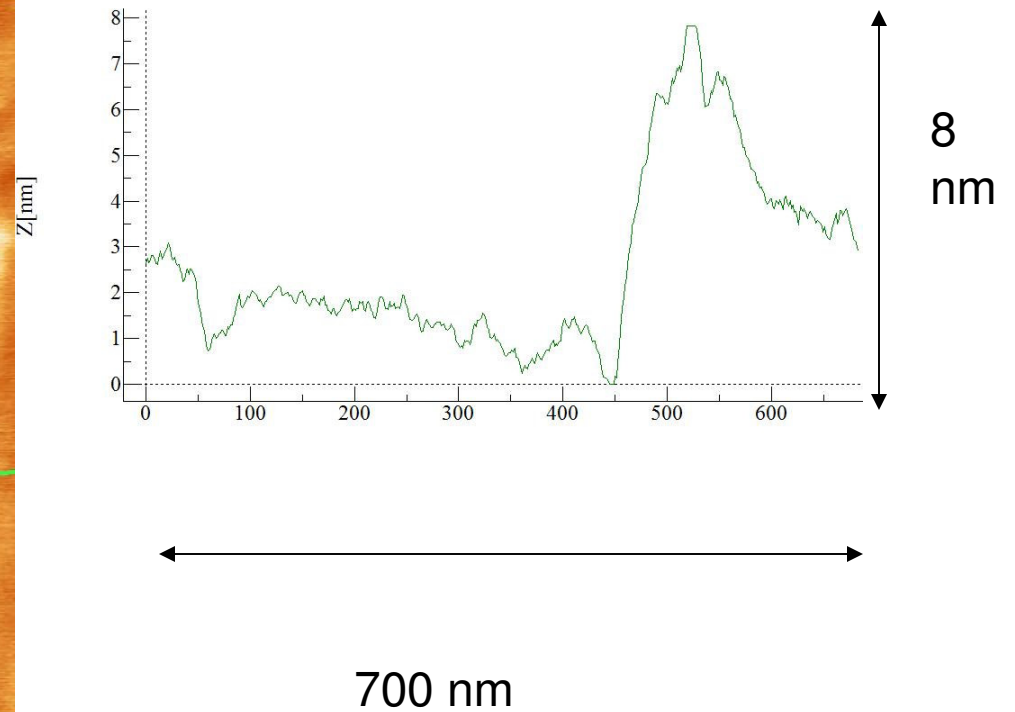
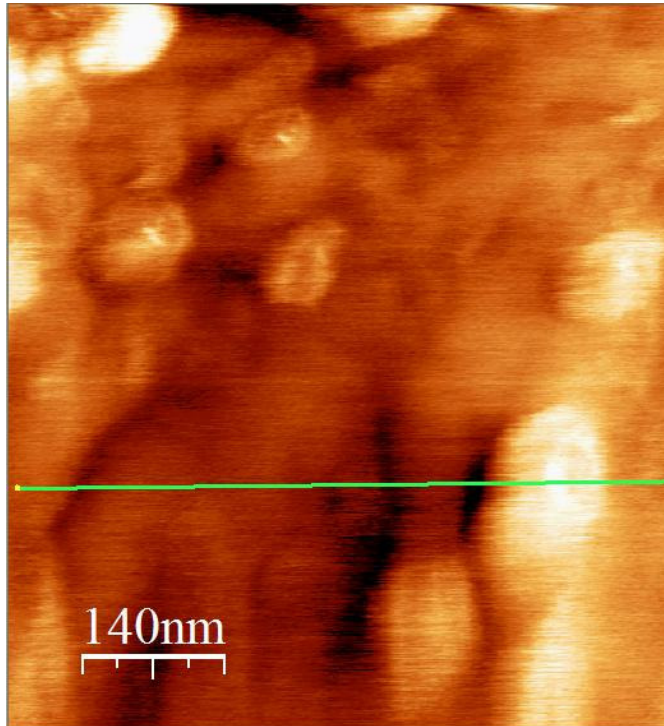
# M vs H at 300 K (MPMS)



# M vs H at 300K (MPMS)



# AFM contact mode



# conclusion

- It is possible to grow epitaxial films of CrO<sub>2</sub> with this technique

# future

- Optimization / control of growth process with respect to e.g film thickness and roughness.
- Using TiO<sub>2</sub> as a substrate instead of sapphire